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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/523,828	02/08/2005	Grant Berent Jacobsen	01435.0207-00000	3990
22852 7590 03/12/2007 FINNEGAN, HENDERSON, FARABOW, GARRETT & DUNNER LLP 901 NEW YORK AVENUE, NW WASHINGTON, DC 20001-4413			EXAMINER	
			KOSLOW, CAROL M	
			ART UNIT	PAPER NUMBER
WASHINGTON	1, 100 20001 1113		01435.0207-00000 3990 EXAMINER KOSLOW, CAROL M	
THE STATE OF THE S	A PERIOD OF RESPONSE	MAIL DATE	DELIVER	Y MODE
SHORTENED STATUTORY	PERIOD OF RESPONSE	WAIL DATE		
3 MON	NTHS	03/12/2007	PAI	PER

Please find below and/or attached an Office communication concerning this application or proceeding.

If NO period for reply is specified above, the maximum statutory period will apply and will expire 6 MONTHS from the mailing date of this communication.

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	Application No.	Applicant(s)	
	10/523,828	JACOBSEN ET AL.	
Office Action Summary	Examiner	Art Unit	
	C. Melissa Koslow	1755	
The MAILING DATE of this communic Period for Reply	cation appears on the cover sheet w	ith the correspondence address	
A SHORTENED STATUTORY PERIOD FO WHICHEVER IS LONGER, FROM THE MA - Extensions of time may be available under the provisions of after SIX (6) MONTHS from the mailing date of this commu If NO period for reply is specified above, the maximum stat Failure to reply within the set or extended period for reply within the set or extended period for reply within the office later than three months after the period patent term adjustment. See 37 CFR 1.704(b).	AILING DATE OF THIS COMMUNION of 37 CFR 1.136(a). In no event, however, may a similation. utory period will apply and will expire SIX (6) MON will, by statute, cause the application to become AF	CATION. reply be timely filed ITHS from the mailing date of this communication BANDONED (35 U.S.C. § 133).	
Status			
1) Responsive to communication(s) filed	b) This action is non-final. or allowance except for formal matt		s
Disposition of Claims			
4) Claim(s) 1-15 is/are pending in the ap 4a) Of the above claim(s) is/are 5) Claim(s) is/are allowed. 6) Claim(s) 1-15 is/are rejected. 7) Claim(s) is/are objected to. 8) Claim(s) are subject to restrict Application Papers , 9) The specification is objected to by the 10) The drawing(s) filed on is/are: Applicant may not request that any object Replacement drawing sheet(s) including the	e withdrawn from consideration. ion and/or election requirement. Examiner. a) accepted or b) objected to bill on abeyar	nce. See 37 CFR 1.85(a).	d).
11)☐ The oath or declaration is objected to	by the Examiner. Note the attached	d Office Action or form PTO-152.	
Priority under 35 U.S.C. § 119		·	
	ocuments have been received. ocuments have been received in A f the priority documents have been al Bureau (PCT Rule 17.2(a)).	application No received in this National Stage	
Attachment(s)			
1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PT 3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date	O-948) Paper No(s	Summary (PTO-413) s)/Mail Date nformal Patent Application 	

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This action is in response to applicants' amendment of 16 August 2006. The amendments to the specification and the claims have overcome the objections to claim 10, the disclosure and the specification. Applicant's arguments with respect to the art rejections have been fully considered but they are not persuasive.

The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.

Claims 1-14 are rejected under 35 U.S.C. 103(a) as being unpatentable over U.S. patent 5,783,512.

This reference teaches a supported catalyst produced by mixing, in a solvent, (a) an ionic activator comprising a cation and an anion containing a moiety having an active hydrogen and (c) an organometal compound; adding the support (d), which is pretreated silica, and then adding (b) a transition metal compound (b) (col. 18, lines 16-25; col. 17, line 65-col. 17, line 15 and col. 17, lines 1-11). Examples 5 and 6 and column 14, line 64 through column 25 teach the pretreated silica can be one that is pretreated with a trialkylaluminum compound, which suggests the use of any trialkylaluminum, such as the claimed triisobutylaluminum. The preferred transition metal compound is a metallocene having a formula which includes the compounds of claim 10 (col. 12, line 57 through column 13, line 20 and examples 1-6). The preferred ionic activator is trialkyl ammonium tris-(pentafluorophenyl)(4-hydroxyphenyl) borate, (col. 8, lines 60-64 and examples). The organometal compound can be an alumoxane, such as that having the formula R₂Al-O(-Al(R)-O)_m-AlR₂, where R can be a C1-10 alkyl and m is 1-50 (col. 13, line 62-col. 14, line 4). This includes the claimed tetraisobutyldialuminoxane. Column 19, lines 3-5 teach the molar ratio of (a) to (c) is 0.05-1000:1, which includes the claimed 20:0.1, or 200:1.

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Product claims with numerical ranges which overlap prior art ranges were held to have been obvious under 35 USC 103. *In re Wertheim* 191 USPQ 90 (CCPA 1976); *In re Malagari* 182 USPQ 549 (CCPA 1974); *In re Fields* 134 USPQ 242 (CCPA 1962); *In re Nehrenberg* 126 USPQ 383 (CCPA 1960). Thus the reference suggests the claimed method of preparing a supported transition metal catalyst.

Column 19, line 13 through column 20, line 67 teaches using the catalyst produced by the above method to polymerize, under polymerization conditions, ethylene, propylene and combinations of ethylene and/or propylene with at least one other α -olefin having 2-8 carbons, such as 1-butene, 1-hexane, 1-octane and 4-methyl-1-pentane. The process can be preformed in the gas, slurry or solution phase. The reference suggests the claimed polymerization process.

Claims 1-15 rejected under 35 U.S.C. 103(a) as being unpatentable over U.S. patent 6,271,165.

This reference teaches a supported catalyst produced by mixing, in a solvent, (a) an ionic activator comprising a cation and an anion containing a moiety having an active hydrogen and (c) an organometal compound; adding the support (d), which is pretreated silica, and then adding (b) a transition metal compound (b) (col. 25, lines 37-40; col. 26, lines 58-col. 27, line 3; col. 29, lines 9-17 and examples). Examples 5, 6, 10-21 and column 24, lines 13-42 teaches the pretreated silica can be one that is pretreated with a trialkylaluminum compound, which suggests the use of any trialkylaluminum, such as the claimed triisobutylaluminum. The preferred transition metal compound is a metallocene having a formula which includes the compounds of claim 10 (col. 17, line 26-col. 21, line 22 and examples). The preferred ionic activator is trialkyl ammonium tris-(pentafluorophenyl)(4-hydroxyphenyl) borate, (col. 13, lines 10-14 and

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examples). The organometal compound can be an alumoxane, such as that having the formula R₂Al-O(-Al(R)-O)_m-AlR₂, where R can be a C1-10 alkyl and m is 1-50 (col. 22, line 65-col. 23, line 10). This includes the claimed tetraisobutyldialuminoxane. Column 29, lines 62-65 teaches the molar ratio of (a) to (c) is 0.05-1000:1, which includes the claimed 20:0.1 or 200:1.Product claims with numerical ranges which overlap prior art ranges were held to have been obvious under 35 USC 103. *In re Wertheim* 191 USPQ 90 (CCPA 1976); *In re Malagari* 182 USPQ 549 (CCPA 1974); *In re Fields* 134 USPQ 242 (CCPA 1962); *In re Nehrenberg* 126 USPQ 383 (CCPA 1960). Thus the reference suggests the claimed method of preparing a supported transition metal catalyst.

Column 30, line 6 through column 31, line 46 and the examples teaches using the catalyst produced by the above method to polymerize, under polymerization conditions, ethylene, propylene and combinations of ethylene and/or propylene with at least one other α -olefin having 2-8 carbons, such as 1-butene, 1-hexane, 1-octane and 4-methyl-1-pentane. The process can be preformed in the gas, slurry or solution phase. Example 12 teaches the gas phase polymerization occurs in a fluidized bed gas phase reactor. The reference suggests the claimed polymerization process.

The declaration is defective since the nationality and address in section are not given and thus, technically, cannot be used to overcome the art rejections. Even if the declaration was not defective, it is insufficient to overcome the rejection of the claim as set forth in the last Office action because: it does not show that the use of aluminoxane gives unexpected results, nor that the order of producing the catalyst is critical. The catalyst preparation methods in the declaration are different from that in specification. For example, the method of making the support in the

declaration is different from that in the specification, the amounts of the borate, aluminum compounds and metallocene in the declaration is different from that in the specification and the solvents are different since the specification teaches the use of cyclohexane and the declaration uses hexane. Given these differences, it cannot be determined what caused the differences shown in the figures. In addition, there is no clear indication that the difference in activity and melt strength between the use of an aluminoxane and an alkylaluminum is in fact unexpected.

The arguments, which are based on the showing in the declaration, are not convincing since there is no showing of unexpected results when using alumoxane nor that the claimed order is critical. The references suggest the claimed process and the use of aluminoxane in the claimed process. The fact the support is optional, the order of mixing is not considered critical and that there are no examples using aluminoxane does not detract form the rejection, since the references do teach embodiments which read on the claimed process. The rejections are maintained.

THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

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Any inquiry concerning this communication or earlier communications from the examiner should be directed to Melissa Koslow whose telephone number is (571) 272-1371. The examiner can normally be reached on Monday-Friday from 8:00 AM to 3:30 PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Jerry Lorengo, can be reached at (571) 272-1233.

The fax number for all official communications is (571) 273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

cmk March 9, 2007 C. Melissa Koslow Primary Examiner Tech. Center 1700 Page 6